

WEST Search History

DATE: Sunday, June 18, 2006

Hide?	Set Name	Query	Hit Count
	<i>DB=PGPB,USPT,USOC,EPAB,JPAB,DWPI; THES=ASSIGNEE; PLUR=YES; OP=ADJ</i>		
<input type="checkbox"/>	L6	L5 not l4	9
<input type="checkbox"/>	L5	L2 and adiabatic	14
<input type="checkbox"/>	L4	L3 and adiabatic	5
<input type="checkbox"/>	L3	L2 and fixed bed	10
<input type="checkbox"/>	L2	L1 and heat exchang\$3 with coolant	62
<input type="checkbox"/>	L1	(synthesis gas or hydrogen near2 carbon oxides) same methanol	3906

END OF SEARCH HISTORY

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NEWS	2		"Ask CAS" for self-help around the clock
NEWS	3	JAN 17	Pre-1988 INPI data added to MARPAT
NEWS	4	FEB 21	STN AnaVist, Version 1.1, lets you share your STN AnaVist visualization results
NEWS	5	FEB 22	The IPC thesaurus added to additional patent databases on STN
NEWS	6	FEB 22	Updates in EPFULL; IPC 8 enhancements added
NEWS	7	FEB 27	New STN AnaVist pricing effective March 1, 2006
NEWS	8	MAR 03	Updates in PATDPA; addition of IPC 8 data without attributes
NEWS	9	MAR 22	EMBASE is now updated on a daily basis
NEWS	10	APR 03	New IPC 8 fields and IPC thesaurus added to PATDPAFULL
NEWS	11	APR 03	Bibliographic data updates resume; new IPC 8 fields and IPC thesaurus added in PCTFULL
NEWS	12	APR 04	STN AnaVist \$500 visualization usage credit offered
NEWS	13	APR 12	LINSPEC, learning database for INSPEC, reloaded and enhanced
NEWS	14	APR 12	Improved structure highlighting in FQHIT and QHIT display in MARPAT
NEWS	15	APR 12	Derwent World Patents Index to be reloaded and enhanced during second quarter; strategies may be affected
NEWS	16	MAY 10	CA/CAPLUS enhanced with 1900-1906 U.S. patent records
NEWS	17	MAY 11	KOREAPAT updates resume
NEWS	18	MAY 19	Derwent World Patents Index to be reloaded and enhanced
NEWS	19	MAY 30	IPC 8 Rolled-up Core codes added to CA/CAPLUS and USPATFULL/USPAT2
NEWS	20	MAY 30	The F-Term thesaurus is now available in CA/CAPLUS
NEWS	21	JUN 02	The first reclassification of IPC codes now complete in INPADOC

NEWS EXPRESS JUNE 16 CURRENT WINDOWS VERSION IS V8.01b, CURRENT
MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP),
AND CURRENT DISCOVER FILE IS DATED 23 MAY 2006.

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=> FIL STNGUIDE

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	0.21	0.21

FILE 'STNGUIDE' ENTERED AT 15:05:06 ON 18 JUN 2006

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=> file caplus

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	0.06	0.27

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FILE COVERS 1907 - 18 Jun 2006 VOL 144 ISS 26

FILE LAST UPDATED: 16 Jun 2006 (20060616/ED)

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=> s (synthesis gas or syngas or hydrogen (2a) carbon oxide?) (p) methanol
1250407 SYNTHESIS
3 SYNTHESISSES
66980 SYNTHESSES
1 SYNTHESSESSES
1287990 SYNTHESIS
(SYNTHESIS OR SYNTHESISSES OR SYNTHESSES OR SYNTHESSESSES)
1503945 GAS
508881 GASES
1685088 GAS
(GAS OR GASES)
16624 SYNTHESIS GAS
(SYNTHESIS (W) GAS)
3979 SYNGAS
14 SYNGASES

3984 SYNGAS
 (SYNGAS OR SYNGASES)
 927455 HYDROGEN
 5776 HYDROGENS
 930689 HYDROGEN
 (HYDROGEN OR HYDROGENS)
 1197876 CARBON
 26137 CARBONS
 1207234 CARBON
 (CARBON OR CARBONS)
 1755903 OXIDE?
 9403 CARBON OXIDE?
 (CARBON(W) OXIDE?)
 193694 METHANOL
 691 METHANOLS
 194059 METHANOL
 (METHANOL OR METHANOLS)
 L1 2543 (SYNTHESIS GAS OR SYNGAS OR HYDROGEN (2A) CARBON OXIDE?) (P)
 METHANOL

=> s l1 and heat exchang?
 1280399 HEAT
 55241 HEATS
 1295144 HEAT
 (HEAT OR HEATS)
 683115 EXCHANG?
 67434 HEAT EXCHANG?
 (HEAT(W) EXCHANG?)
 L2 103 L1 AND HEAT EXCHANG?

=> s l2 and fixed bed
 228818 FIXED
 1 FIXEDS
 228819 FIXED
 (FIXED OR FIXEDS)
 164690 BED
 65860 BEDS
 188534 BED
 (BED OR BEDS)
 19967 FIXED BED
 (FIXED(W) BED)
 L3 3 L2 AND FIXED BED

=> s l2 and coolant
 35231 COOLANT
 13991 COOLANTS
 40735 COOLANT
 (COOLANT OR COOLANTS)
 L4 3 L2 AND COOLANT

=> s l3 or l4
 L5 5 L3 OR L4

=> d l5 ibib ab 1-5

L5 ANSWER 1 OF 5 CAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2005:558822 CAPLUS
 DOCUMENT NUMBER: 143:155677
 TITLE: Method for catalytic synthesis of dimethyl ether in
 combined bed reactor
 INVENTOR(S): Ying, Weiyong; Fang, Dingye; Zhang, Haitao; Liu,
 Dianhua; Cao, Fahai
 PATENT ASSIGNEE(S): East China University of Science and Technology, Peop.
 Rep. China

SOURCE: Faming Zhuanli Shenqing Gongkai Shuomingshu, 9 pp.
 CODEN: CNXXEV
 DOCUMENT TYPE: Patent
 LANGUAGE: Chinese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1413974	A	20030430	CN 2002-136724	20020829
PRIORITY APPLN. INFO.:			CN 2002-136724	20020829

AB Di-Me ether is synthesized by reaction of **syngas** in (medical) liquid paraffin in the presence of catalyst in a three-phase slurry-bed/**fixed-bed** combined reactor at 220-280° and 3.0-7.0 MPa. The **syngas** is prepared from natural gas or coal. The catalyst is composed of Cu series **methanol** synthesis catalyst and modified mol. sieve (ratio 0.4-2.0:1). The combined reactor consists of a reactor body that is divided into a three-phase slurry bed section and a **fixed bed** section, a **heat exchanger**, a gas distributing unit, a gas pocket, and a separator between reactor sections.

L5 ANSWER 2 OF 5 CAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 2004:633898 CAPLUS
 DOCUMENT NUMBER: 141:158961
 TITLE: Hydrogenation process for **methanol** manufacture from **synthesis gas**
 INVENTOR(S): Fitzpatrick, Terence James
 PATENT ASSIGNEE(S): Johnson Matthey Plc, UK
 SOURCE: PCT Int. Appl., 21 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2004065341	A1	20040805	WO 2004-GB75	20040112
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ				
AU 2004205368	A1	20040805	AU 2004-205368	20040112
CN 1741978	A	20060301	CN 2004-80002577	20040112
US 2006074133	A1	20060406	US 2005-542819	20050720
PRIORITY APPLN. INFO.:			GB 2003-1323	A 20030121
			WO 2004-GB75	W 20040112

AB **Methanol** is synthesized from pre-heated **methanol synthesis gas** in one or more adiabatic synthesis stages with cooling of the resultant gas after each stage. Further **methanol** synthesis is then effected on the resultant, partially reacted **synthesis gas** in a bed of synthesis catalyst cooled by means of a **coolant** flowing concurrently through tubes disposed in the catalyst bed. After cooling, **methanol** is separated from the unreacted gas. Part of the unreacted gas is combined with make-up gas and used as the **coolant** fed to the aforesaid tubes, thus producing the pre-heated **synthesis gas** to be fed to the adiabatic synthesis stages. A process flow diagram is presented.

REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 3 OF 5 CAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 2003:8371 CAPLUS

DOCUMENT NUMBER: 138:340592
 TITLE: Synthesis Gas Production in a Forced Unsteady-State Reactor Network
 AUTHOR(S): Fissore, Davide; Barresi, Antonello A.; Baldi, Giancarlo
 CORPORATE SOURCE: Dipartimento di Scienza dei Materiali ed Ingegneria Chimica, Politecnico di Torino, Turin, 10129, Italy
 SOURCE: Industrial & Engineering Chemistry Research (2003), 42(12), 2489-2495
 CODEN: IECRED; ISSN: 0888-5885
 PUBLISHER: American Chemical Society
 DOCUMENT TYPE: Journal
 LANGUAGE: English

AB The feasibility of producing **synthesis gas** by the combination of partial oxidation and steam reforming of natural gas on a Pt-based catalyst in forced unsteady-state catalytic reactors was considered by numerical simulations. A network of three reactors with periodical change of the feed position was investigated as an alternative to the well-investigated reverse-flow reactor: these modes of reactor operation may lead to lower **syngas** manufacturing costs than the conventional unidirectional **fixed-bed** reactor because external **heat exchangers** are no longer required. A cyclic steady-state condition and autothermal behavior can be obtained by feeding low-temperature reactants. The influence of the main operating parameters (inlet temperature, switching time, inlet flow rate, and composition) on the performance of the device was investigated, proving that the network can be competitive with traditional technologies, allowing for higher reactant conversion and product selectivity. The possibility of tailoring the H₂/CO ratio to the value required for the production of **methanol** or Fischer-Tropsch synthesis was addressed.

REFERENCE COUNT: 22 THERE ARE 22 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 4 OF 5 CAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1999:753188 CAPLUS
 DOCUMENT NUMBER: 131:338608
 TITLE: **Methanol** manufacture from **synthesis gas** made by steam reforming of hydrocarbons using indirect **heat exchange**
 INVENTOR(S): Fitzpatrick, Terence James
 PATENT ASSIGNEE(S): Imperial Chemical Industries PLC, UK
 SOURCE: PCT Int. Appl., 19 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 2
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 9959945	A1	19991125	WO 1999-GB1335	19990429
W: AU, BR, CA, GE, ID, JP, KR, MX, NO, UA, US, UZ, ZA, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE				
CA 2330298	AA	19991125	CA 1999-2330298	19990429
AU 9937207	A1	19991206	AU 1999-37207	19990429
AU 740997	B2	20011122		
EP 1080059	A1	20010307	EP 1999-919408	19990429
EP 1080059	B1	20040218		
R: DE, GB, NL				
JP 2002515468	T2	20020528	JP 2000-549565	19990429
US 6387963	B1	20020514	US 2000-714486	20001117

US 6433029 B1 20020813 US 2000-714218 20001117
 PRIORITY APPLN. INFO.: GB 1998-10700 A 19980520
 GB 1998-11355 A 19980528
 GB 1999-4649 A 19990302
 WO 1999-GB1335 W 19990429
 WO 1999-GB1344 A1 19990429

AB Methanol is manufactured in high yield and selectivity in a synthesis loop having at least two synthesis stages where methanol is prepared from recycled, unreacted gas, optionally together with part of the synthesis gas, in one or more synthesis stages to give a stream of reacted gas, synthesis gas is then added and prior to separation of the methanol, a further amount of methanol is synthesized from the resultant mixture in one or more further synthesis stages, with at least the final synthesis stage of the loop being effected via indirect heat exchange with pressurized water as the coolant. Preferably the pressurized hot water from the final synthesis stage of the loop is employed to saturate a hydrocarbon feedstock (e.g., natural gas) from which the synthesis gas is produced by steam reforming. Process flow diagrams are presented.

REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 5 OF 5 CAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1989:635591 CAPLUS

DOCUMENT NUMBER: 111:235591

TITLE: Process and catalyst for the manufacture of methanol from synthesis gas

INVENTOR(S): Sie, Swan Tiong; Van Dijk, Arjan

PATENT ASSIGNEE(S): Shell Internationale Research Maatschappij B. V., Neth.

SOURCE: Eur. Pat. Appl., 5 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 322989	A2	19890705	EP 1988-203039	19881229
EP 322989	A3	19900207		
R: BE, DE, ES, FR, GB, IT, NL				
JP 01233241	A2	19890919	JP 1988-329577	19881228
ZA 8809668	A	19891129	ZA 1988-9668	19881228
AU 8827563	A1	19890706	AU 1988-27563	19881229
AU 605655	B2	19910117		

PRIORITY APPLN. INFO.: GB 1987-30280 A 19871230

AB MeOH is prepared by the hydrogenation of CO in the presence of a catalyst system prepared by combining a Ni salt with an alkali metal alcoholate or an alkaline earth metal alcoholate. An inert liquid coolant, immiscible with MeOH, at 0-70°, is injected into the reaction liquid phase and serves as a coolant. The coolant (e.g., n-pentane) does not deactivate the catalyst and overcomes the problems of large indirect heat exchange surface area requirements by being part of the reaction mixture. The coolant is removed with the MeOH product by vaporization and recovered from the MeOH by phase separation. In this manner, synthesis gas (H/CO volume ratio 2) was converted into MeOH at 120°/15 bar in the presence of a Ni formate-NaH-tert-pentyl alc.-diglyme catalyst system and n-pentane (coolant liquid hourly space velocity 1000 kg/m³-h), producing a 2-phase product (the upper phase comprising 98% n-pentane and 2% MeOH; the lower phase comprising MeOH 85, n-pentane 5, and H₂O 10%) the phases separated, the MeOH removed from the lower phase by distillation, and the n-pentane